# Rigid Molecular Architectures That Comprise a 1,3,5-Trisubstituted Benzene Core and Three Oligoaryleneethynylene Arms: Light-Emitting Characteristics and $\pi$ Conjugation between the Arms 

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Rigid molecular architectures which consist of a trivalent core and three $\pi$ extended arms (so-called star-shaped molecules) are currently attracting considerable attention because of potential application as devices, ${ }^{1}$ such as light-emitting diodes (LEDs), field effect transistors (FETs), and nonlinear optics (NLO). However, with respect to the light-emitting characteristics associated with LEDs, only a few papers (dealing with the light-emitters with $\Phi_{\mathrm{f}}$ $<0.86)$ appear in the literature. ${ }^{2}$ Apart from this, it should be interesting to examine if $\pi$ conjugation occurs between the arms across the $1,3,5$-trisubstituted benzene core of star-shaped molecules, though such an interaction is reported for condensed polycyclic systems. ${ }^{3}$ Thus we have investigated the light-emitting characteristics and occurrence of $\pi$ conjugation between the arms of star-shaped rigid molecules that comprise a $1,3,5$-triethynylbenzene core and methoxy group-substituted oligo( $p$-phenyleneethynylene) arms. Consequently, we achieved the ultimate goal ( $\Phi_{\mathrm{f}} \approx$ $1.0, \log \epsilon>5$ ) for organic molecules with respect to light-emitting ability by integrating the dimethoxyphenylacetylene unit with low $\Phi_{\mathrm{f}}$ ( 0.29 in cyclohexane), not the fluorene unit with high $\Phi_{\mathrm{f}}(0.74$ in cyclohexane). Furthermore, we have found that $\pi$ conjugation occurs between the arms.


1


2


3

$4(n=0)$
$6(n=1)$
$6(n=1)$
$8(n=2)$
$8(n=2)$

$7(n=1)$
$9(n=2)$

In regard to the star-shaped molecules, we considered 1 (the simplest star-shaped hydrocarbon system), $\mathbf{2}$ (cyano-substituted $\mathbf{1}$ at each terminal), $\mathbf{3}$ (methoxy-substituted $\mathbf{1}$ at each terminal), 4, 6, and $\mathbf{8}$ (homologues containing zero, one, and two dimethoxyphenyleneethynylene units and a monomethoxyphenyl terminal at each arm) (abbreviated as MMPT homologues), and 5, 7, and 9 (homologues containing zero, one, and two dimethoxyphenyleneethynylene units and dimethoxyphenyl terminal at each arm) (abbreviated as DMPT homologues). Synthesis of $\mathbf{1 - 1 0}$ was achieved by a $\mathrm{Pd} \mathrm{C}-\mathrm{C}$ cross-coupling reaction (see Supporting Information).

The photophysical data ${ }^{4}$ of star-shaped molecules 1-9 together with radiative rate constant $\left(k_{\mathrm{r}}\right)$, radiationless rate constant $\left(k_{\mathrm{d}}\right), k_{\mathrm{r}} /$

Table 1. Photophysical Data of Star-Shaped Molecules 1-9 in $\mathrm{CHCl}_{3}{ }^{\text {a }}$

|  |  | $\lambda_{\text {em }}$ <br> $(\mathrm{nm})$ | $\log \epsilon$ | $\lambda_{\text {abs }}$ <br> $(\mathrm{nm})$ | $\tau$ <br> $(\mathrm{ns})$ | $k_{\mathrm{r}}$ <br> $\left(\mathrm{s}^{-1}\right)$ | $k_{\mathrm{d}}$ <br> $\left(\mathrm{s}^{-1}\right)$ | $k_{\mathrm{f}} / k_{\mathrm{d}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| compound | $\Phi_{\mathrm{f}}{ }^{\text {b }}$ |  |  |  |  |  |  |  |
| $\mathbf{1}$ | 0.15 | 353 | 4.93 | 305 | 1.17 | $1.28 \times 10^{8}$ | $7.23 \times 10^{8}$ | 0.18 |
| $\mathbf{2}$ | 0.14 | 357 | 5.11 | 320 | 0.78 | $1.80 \times 10^{8}$ | $1.11 \times 10^{9}$ | 0.16 |
| $\mathbf{3}$ | 0.23 | 360 | 4.93 | 316 | 1.17 | $1.96 \times 10^{8}$ | $6.55 \times 10^{8}$ | 0.30 |
| $\mathbf{4}$ | 0.24 | 359 | 4.77 | 314 | 1.70 | $1.41 \times 10^{8}$ | $4.48 \times 10^{8}$ | 0.32 |
| $\mathbf{5}$ | 0.46 | 384 | 4.81 | 334 | 1.55 | $2.97 \times 10^{8}$ | $3.49 \times 10^{8}$ | 0.85 |
| $\mathbf{6}$ | 0.83 | 406 | 5.02 | 377 | 0.95 | $8.69 \times 10^{8}$ | $1.78 \times 10^{8}$ | 4.88 |
| $\mathbf{7}$ | 0.85 | 409 | 5.11 | 380 | 0.78 | $1.10 \times 10^{9}$ | $1.93 \times 10^{8}$ | 5.67 |
| $\mathbf{8}$ | 0.97 | 433 | 5.1 | 405 | 0.78 | $1.25 \times 10^{9}$ | $3.86 \times 10^{7}$ | 32.33 |
| $\mathbf{9}$ | 0.98 | 464 | 5.29 | 426 | 0.51 | $1.93 \times 10^{9}$ | $3.93 \times 10^{7}$ | 49.00 |

[^0] yield is calculated relative to quinine ( $\Phi_{\mathrm{f}}=0.55$ in $0.1 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}$ ).
$k_{\mathrm{d}}$ (a measure for emissivity) and emission lifetime $(\tau)$ are summarized in Table 1. Since the $k_{\mathrm{r}}$ and $k_{\mathrm{d}}$ are related to the corresponding emission quantum yields and lifetimes by $\Phi_{\mathrm{f}}=k_{\mathrm{r}}$ $\times \tau$ and $k_{\mathrm{r}}+k_{\mathrm{d}}=\tau^{-1}$, it is possible to calculate the values of $k_{\mathrm{r}}$ and $k_{\mathrm{d}}$ wherever quantum yield and lifetime data are available. ${ }^{5}$

As demonstrated in Table 1, the introduction of a methoxy (donor) group into the benzene rings of each arm of the simplest star-shaped hydrocarbon system $\mathbf{1}$ brings about a slight red shift of the fluorescence emission maximum ( $\lambda_{\mathrm{em}}$ ) and an increase in quantum yield ( $\Phi_{\mathrm{f}}$ ) (3, 4), though the introduction of a cyano (acceptor) group does not improve the $\Phi_{\mathrm{f}}$ value (2). Although the emission efficiency ( $\Phi_{\mathrm{f}}$ ) increases with increasing dimethoxyphenyleneethynylene unit in each arm, the light-emitting characteristics of star-shaped molecules ( $\mathbf{4 - 9}$ ) can be better understood by dividing into two homologous series, MMPT (4, 6, and $\mathbf{8})$ and $\operatorname{DMPT}(\mathbf{5}, \mathbf{7}$, and $\mathbf{9}$ ). For both series, the values of $\Phi_{\mathrm{f}}, \lambda_{\mathrm{em}}, \log \epsilon$, and $\lambda_{\text {abs }}$ increase regularly with a number of dimethoxyphenyleneethynylene units. Thus we succeeded in creating very intense violet-blue ( $\mathbf{8}, \Phi_{\mathrm{f}}=$ $0.97, \log \epsilon=5.11)$ and blue ( $9, \Phi_{\mathrm{f}}=0.98, \log \epsilon=5.29$ ) bright light-emitters, and thereby we achieved the ultimate goal ( $\Phi_{\mathrm{f}} \approx$ $1.0, \log \epsilon>5$ ) for organic molecules with respect to light-emitting ability.
We examined the relations of $k_{\mathrm{r}}$ and of $k_{\mathrm{d}}$ with the number of dimethoxyphenyleneethynylene units excluding the terminal aryleneethynylene groups ( $\mathbf{3 n}$, a measure for $\pi$ extension). Consequently, as seen in Figure 1, we found ${ }^{6}$ linear relationships between $k_{\mathrm{r}}$ and 3n (slope $=1.85 \times 10^{8}, r^{2}=0.9684$ ) and $k_{\mathrm{d}}$ and $\mathbf{3 n}$ (slope $=$ $-0.68 \times 10^{8}, r^{2}=0.9672$ ) in the MMPT series, as well as between $k_{\mathrm{r}}$ and 3 n (slope $=2.79 \times 10^{8}, r^{2}=0.9995$ ) and $k_{\mathrm{d}}$ and 3 n (slope $\left.=-0.51 \times 10^{8}, r^{2}=1.0000\right)$ in the DMPT series. For both series, $k_{\mathrm{r}}$ values increase and $k_{\mathrm{d}}$ values decrease with $\mathbf{3 n}$, indicating that emission efficiency of the homologues in both series increases with the number of dimethoxyphenyleneethynylene units. However, from slopes for plots of $k_{\mathrm{r}}$ versus $\mathbf{3 n}$ and $k_{\mathrm{d}}$ versus $\mathbf{3 n}$, the DMPT


Figure 1. The relationship of $k_{\mathrm{r}}$ and $k_{\mathrm{d}}$ to $\mathbf{3 n}$ in MMPT and DMPT homologues (blue line for MMPT homologues, red line for DMPT homologues).

Table 2. Photophysical Data of Star-Shaped System (9), Banana-Shaped System (10), and Rod-Shaped System (11) in $\mathrm{CHCl}_{3}{ }^{\text {a }}$

| compound | $\Phi_{\mathrm{f}}{ }^{b}$ | $\lambda_{\text {em }}$ <br> $(\mathrm{nm})$ | $\log \epsilon$ | $\lambda_{\text {abs }}$ <br> $(\mathrm{nm})$ | $\Delta \lambda_{\text {abs }}$ <br> $(\mathrm{nm})$ | $\epsilon / \epsilon_{11}$ | $k_{\mathrm{r}}$ <br> $\left(\mathrm{s}^{-1}\right)$ | $k_{\mathrm{d}}$ <br> $\left(\mathrm{s}^{-1}\right)$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{9}$ | 0.98 | 464 | 5.29 | 426 | 36 | 3.50 | $1.93 \times 10^{9}$ | $3.93 \times 10^{7}$ |
| $\mathbf{1 0}$ | 0.90 | 432 | 4.96 | 402 | 12 | 1.59 | $8.02 \times 10^{8}$ | $8.92 \times 10^{7}$ |
| $\mathbf{1 1}$ | 0.81 | 430 | 4.75 | 390 | 0 | 1.00 | $4.55 \times 10^{8}$ | $1.07 \times 10^{8}$ |

${ }^{a}$ All spectra were measured for $10^{-7} \mathrm{M}$ solution at $295 \mathrm{~K} .{ }^{b}$ Quantum yield is calculated relative to quinine ( $\Phi_{\mathrm{f}}=0.55$ in $0.1 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}$ ).

homologues are favored over the MMPT homologues with respect to the emission characteristics. The emissive superiority of the DMPT homologues to the MMPT homologues might be explained by movability (number) of the dipolar dimethoxyphenylacetylene unit (donor, OMe ; acceptor, $\mathrm{C} \equiv \mathrm{C}$ ) in the excited singlet state as suggested in the previous paper. ${ }^{7}$

To know if any $\pi$ conjugation occurs between the arms of starshaped molecules, we have examined the $\Delta \lambda_{\text {abs }}$ and $\epsilon / \epsilon_{11}$ values for 1,3,5-trisubstituted benzene ( 9 , star shape), 1,3-disubstituted benzene (10, banana shape), and monosubstituted benzene (11, rod shape) ${ }^{8}$ with the trimeric(dimethoxyphenyleneethynylene) group. The results are summarized in Table 2.

As seen in Table 2, the values of $\Delta \lambda_{\text {abs }}$ and $\epsilon / \epsilon_{11}$ evidently increase in the order of $\mathbf{9}($ star shape $)>\mathbf{1 0}$ (banana shape) $>\mathbf{1 1}$ (rod shape). Difference in $\lambda_{\text {abs }}$ between 9 and 11 becomes 36 nm , and $\epsilon_{9} / \epsilon_{11}$ becomes 3.50. These data clearly indicate an extension of the $\pi$ conjugation between the arms. Although an increase in $\Delta \lambda_{\text {abs }}$ is reported for condensed polycyclic systems, ${ }^{3}$ the star-shape molecule (9) seems to be a more straightforward example for meta conjugation.

The solvent dependency on the emission characteristics of $\mathbf{8}$ and $\mathbf{9}$ is contrasting (see Supporting Information). The $\Phi_{\mathrm{f}}$ and $\lambda_{\mathrm{em}}$ values of 9 are remarkably altered with a change in solvent polarity, while little solvent dependency on $\Phi_{\mathrm{f}}$ and $\lambda_{\mathrm{em}}$ values of $\mathbf{8}$ is observed, ${ }^{9}$ which seems to be explained in terms of a marked disparity between difference density distribution in the excited singlet and the ground state (see Supporting Information).

In conclusion, we have synthesized the star-shaped architectures that comprise a 1,3,5-triethynylbenzene core and methoxy groupsubstituted oligo( $p$-phenyleneethynylene) arms. We achieved the ultimate goal $\left(\Phi_{\mathrm{f}} \approx 1.0, \log \epsilon>5\right)$ for organic molecules with respect to light-emitting ability by creating very intense violet-blue $\left(\mathbf{8}, \Phi_{\mathrm{f}}=0.97, \log \epsilon=5.11\right)$ and blue $\left(9, \Phi_{\mathrm{f}}=0.98, \log \epsilon=5.29\right)$ bright light-emitters. We have found that $\pi$ conjugation occurs between the arms of 9 despite the meta-substituted system. A linear relationship of $k_{\mathrm{r}}$ (with positive slope) and of $k_{\mathrm{d}}$ (with negative slope) to the number of dimethoxyphenyleneethynylene units (3n) for MMPT (4, 6, and 8) and DMPT (5, 7, and 9) homologues and the contrasting solvent effect on $\lambda_{\text {em }}$ of $\mathbf{8}$ and $\mathbf{9}$ are also interesting findings. It is noted that $\Phi_{\mathrm{f}}, \lambda_{\mathrm{em}}, \lambda_{\mathrm{abs}}, k_{\mathrm{r}}, k_{\mathrm{d}}$, and solvent effect are controlled by the number (one or two) of the methoxy groups on the terminal benzene rings.

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Supporting Information Available: Synthesis, NMR data, HR MS data, fluorescence spectra for $\mathbf{6}, \mathbf{7}, \mathbf{8}$, and $\mathbf{9}$, solvent effect on spectra of $\mathbf{8}$ and $\mathbf{9}$, and MO and MM2 calculation for $\mathbf{8}$ and 9 . This material is available free of charge via the Internet at http://pubs.acs.org.

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(9) We have confirmed that the $\Phi_{\mathrm{f}}$ and $\lambda_{\mathrm{em}}$ values of $\mathbf{8}$ and $\mathbf{9}$ in polar solvents are not so altered with change in concentration from $10^{-6}$ to $10^{-8} \mathrm{M}$. JA057751R


[^0]:    ${ }^{a}$ All spectra were measured for $10^{-7} \mathrm{M}$ solution at $295 \mathrm{~K} .{ }^{b}$ Quantum

